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A REVIEW ON THERMAL ENERGY STORAGE USING PHASE CHANGE MATERIAL

Gaikwad Kunal¹, Sinha Shubham², Gurav Shubham³, Kiran Shivade⁴

^{1,2,3}TE MECHANICAL SCHOLAR, BVCOE & RI, Nashik (Pune University)

⁴Assistant Professor, Mechanical Dept. BVCOE&RI, Nashik, (India)

ABSTRACT

Thermal energy storage (TES) systems provide several alternatives for capability energy use and conservation. Phase change materials (PCMs) for TES are materials supplying thermal regulation at particular phase change temperatures by absorbing and emitting the heat of the medium. TES in general and PCMs in particular, have been a main topic in research for the last 30 years, but although the information is quantitatively enormous, it is also spread widely in the literature, and difficult to find. PCMs absorb energy during the heating task as phase change takes place and release energy to the environment in the phase change range during a reverse cooling task. PCMs possesses the ability of latent thermal energy change their state with a certain temperature. PCMs for TES are generally solid-liquid phase change materials and therefore they need encapsulation. TES systems using PCMs as a storage medium offers advantages such as high TES capacity, small unit size and isothermal behavior during charging and discharging when compared to the prudent TES.

Keywords: Phase Change Material (PCM); Thermal Energy Storage (TES).

I. INTRODUCTION

Thermal energy storage (TES) is thought because the transient holding of thermal energy within the sort of hot or cold physical matter for when usage. Energy demands disagree on a day, weekly and seasonal based mostly. These demands are often matched with the assistance of TES management that operate synergistically, and to distribute with the storage of energy by cooling, heating, melting, natural process or evaporate a cloth and therefore the thermal energy becomes valid once the task is opposite. TES may be a important technology in systems involving renewable energies moreover as alternative energy resources because it will build their operation additional not overwhelming additional, significantly by bridging the amount between periods once energy is resulted and periods once it's required. That is, TES is useful for equalization between the availability and demand of energy [1, 2]. TES systems have the potential for growing the effective use of thermal energy instrumentality and for facilitating large-scale fuel commutating [2]. The selection of a TES system for a specific application depends on several factors, among storage length, economics, provide and utilization temperature necessities, storage power, heat losses and on the market area [3].

The main varieties of TES ar prudent and latent. Prudent TES systems store energy by dynamic the temperature of the storage means that, which may be water, brine, rock, soil, etc. Latent TES systems store energy through phase transition, e.g., cold storage water/ice and warmth storage by melting paraffin waxes. Latent TES units are typically smaller than prudent storage units. Additional compact TES are often achieved supported storages that resort chemical reactions [1].

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II.LITERATURE SURVEY

Karthikeyan.S[1]"Thermal energy storage in phase change material".(TES) system provide several alertanitives for capability energy use and conservation (TES) in general and PCM's in particular, have been a main topic in research for the last thirty years but although the information is quantitatively enormous, it is also spread widely in the literature and difficult to find. PCM's absorb energy during the heating task as a phase change take place and release energy to the environment in the phase change range during a reverse cooling task.

Ravikumar.M[2],"Use of PCM material in development of building". PCM exhibit thermodynamic property of storing large amount of latent heat during its phase change. PCM solidifies on drop of ambient temperature giving off its latent heat of fusion. PCM are implemented in gypsum wall boards, plasters, texture finish due to its thermal storage application. Super cooling influence the performance of PCM.

Ravikumar M., Srinivasan Dr.Pss[3],"Natural cooling of building using phase change material". Electrical energy consumption varies significantly during the day and night according to the demand of industrial, commercial and residential. Recent discussion on topic like global warming and heat waves have brought attention once again to energy capability cooling system. Utilizing renewable energy sources. Thermal storage play an vital role in building energy conservation.

Zalba B., Marın J.M., Cabeza L.F., Mehling H[4], "Review on thermal energy storage with phase change: materials, heat transfer analysis and applications". Stusies has reveal that this energy capability structure can experience problems with thermal discomfort due to elevated Indore air temperature in warm weather. This problem is toincorporate PCM into the construction fabric, in order to increase the effective heat storage capacity of small buildings.

Khudhair A.M., Farid M.M[5],"A review on energy conservation in building applications with thermal storage by latent heat using phase change materials, Energy Conversion and Management". The use of novel building materials containing active thermal components.(eg. Subventing radiant barriers) would be an ultimate step in acchieving heat and cooling energy saving from technological building envelope improvement.

III.TECHNOLOGY

A complete TES task engages a minimum of 3 steps: charging, storing and discharging. In sensible systems a number of the steps could happens at constant time (for example charging and storing) every and every} step could occur over once in each storage cycle. In figure one is illustrated a straightforward storage cycle, during which the 3 steps ar shown separate. Wherever the warmth Ql is penetrate and is positive in price for a chilly thermal storage. If it's discharged, it'll be toward the atmosphere and Ql are negative. The warmth proceed is illustrated for the storing a task, however will occur altogether 3 task [3]. In figure a pair of is bestowed the rise of internal energy, once energy within the kind of heat is extra to a fabric. The well-known outcome is a rise in temperature (prudent TES) or amendment of section (latent TES). Beginning with AN initial solid state at purpose O, a heat addition to the fabric initial causes prudent heating of the solid (region O–A), followed by a solid-to-liquid natural action (region A–B), a prudent heating of the liquid (region B–C), a liquid-to-vapour.

Phase change (region C–D), and a prudent heating of the vapour (region D–E). The overall quantity of warmth may be written within the following formula [4]:

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$$Q = m \cdot \left[\int_{TO}^{TA} C_{ps}(T) dT + q_t + \int_{TB}^{TC} C_{pl}(T) dT + q_i + \int_{TD}^{TE} C_{pv}(T) dT \right]$$

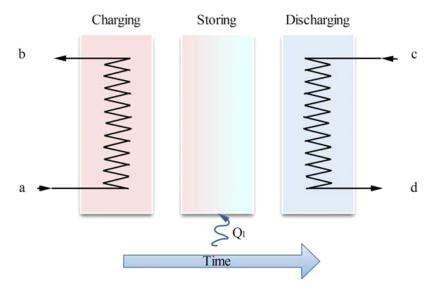


Figure 1. The three taskes in a general TES system

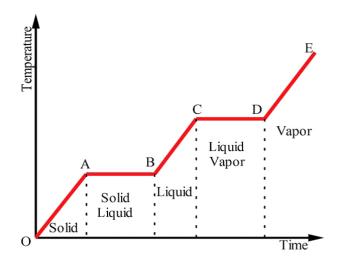


Figure 2. Temperature-time diagram for the heating of a material

Latent heat storage is one amongst the foremost capability ways that of storing thermal energy [5]. In latent TES systems, energy is keep throughout the action (e.g. melting, evaporating and Thermal Energy Storage with action Material Lavinia Gabriela SOCACIU 78crystallization). Thanks to the particular heat of a typical medium and therefore the high physical property amendment throughout action, the heat amendment is typically larger than the prudent heat amendment for a given system size [1]. Not like the prudent heat storage methodology, the heat storage methodology provides abundant higher storage density, with a brief temperature distinction between storing and emotional heat. Every material absorbs heat throughout heating task whereas its

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temperature is rising oftentimes. the warmth keep within the material is discharged into the system through a reverse cooling task. Throughout the cooling task, the fabric temperature decreases endlessly [5]. The keep energy throughout a latent storage task are often evaluated as: Q=m•L (2) wherever m denotes the mass and L is that the specific heat of the PCM (Phase amendment Material) [1]. Latent TES systems store energy in PCMs, with the thermal energy keep once the fabric changes part, typically from a solid to liquid (for example: energy is needed to convert ice to water, to vary water to steam and to soften paraffin wax). the foremost common example of latent TES is that the conversion of water to ice. Cooling systems incorporating ice storage have a definite size advantage over equivalent-capacity chilled-water units became of the comparatively great deal of energy that's keep through the action [3]. For minus (cold) temperature, PCMs (i.e. ice), the liquid to solid (freezing) amendment absorbs energy and therefore the solid to liquid amendment releases that absorbed energy. On the opposite hand, for positive (hot) temperature PCMs, the solid to liquid amendment absorbs energy and therefore the liquid to solid amendment releases that absorbed energy, and will therefore at constant temperatures. In every case, the number of energy absorbed and discharged is termed as heat [6]. action task of PCM from solid to liquid and the other way around is schematically shown in figure three.

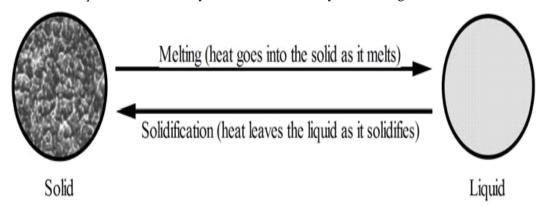


Figure 3. Schematic representation of phase change task

The large heat transfer throughout the melting task furthermore because the crystallization task while not important natural action makes PCM fascinating as a supply of warmth storage material in sensible applications. Once temperature will increase, the PCM microcapsules absorbed heat and storing this energy within the liquefied activity materials. Once the temperature falls, the PCM microcapsules unleash this hold on heat and consequently PCM solidify [5]. The energy needed to cause these changes is called the warmth of fusion at the purpose freezingpoint temperature} and also the heat of vaporization at the boiling point. The particular heat of fusion or vaporization and also the temperature at that the activity happens area unit important in style section. PCMs area unit either prepackaged in specialized containers such as: tubes, having little dept. panels, plastic bags; or contained in standard building parts such as: wall board and ceiling; or encapsulated as self-contained parts [1, 3]. The aim of this analysis paper was to produce a compilation of sensible

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Latent Thermal Energy Storage Gas-Liquid Solid-Gas Solid-Liquid Solid-Solid **Organics** Inorganics **Eutetics Single Temperature** Mixtures Temperature Interval Mixtures Temperature Interval **Eutetics Single Temperature** Paraffins (alkanes mixtures) Fatty acids Hydrated salts Commercial grade Analytical grade

Chart no.1 classification of energy storage

Melting Temperature | Heat of Fusion PCM Name [°C] [kJ/kg] 19 CH3(CH2)16COO(CH2)3CH3 Butyl stearate 140 CH₃(CH₂)₁₁OH 1-dodecanol 26 200 $CH_3(CH_2)_{12}OH$ 1-tetradecanol 38 205 20-60 $CH_3(CH_2)_n(CH3...$ Paraffin 200 45%CH3(CH2)8COOH 45/55 21 143 capric-lauric acid 55%CH₃(CH₂)₁₀COOH 19 CH₃(CH₂)₁₂COOC₃H₇ Propyl palminate 186

Table 2. Organic PCMs (typical values)

IV. METHODOLOGY

In our topic, initially we have taken different phase change material and doing experimental analysis and that reading data is calibrated with standard PCM material and then it is observe that PCM material are having high thermal storage capacity

V. FUTURE SCOPE

PCM materials will be use as a cooling material.

PCM materials use in building material to store thermal energy, it can be use as a energy conservation material.

VI. CONCLUSION

The incorporation of PCMs into building parts takes the advantage of latent TES for added energy savings. the event of anabolic building could be a answer to the on-going go after energy conservation, and conjointly to up the indoor surroundings within which individuals work and live. In terms of thermal comfort, it's envisaged that the indoor surroundings of a building that uses PCM construction materials can have considerably lower mean

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beaming temperatures and additional thermal stability, having less probability of heating and fewer temperature fluctuations. Thermal enhancements in a very building thanks to the inclusion of PCMs rely upon the sort of PCM, the melting temperature, the proportion of PCM mixed with standard material, the climate, style and orientation of the development of the building. The improvement of those parameters is prime to demonstrate the probabilities of success of the PCMs in building materials.

REFERENCES

- [1]. Karthikeyan S., Muthu Saravanan S., Prashanth R., Energy conservation through phase change material based thermal energy storage system-a project report, Anna UniversityChennai, 2006.
- [2]. Ravikumar M., Srinivasan Dr.Pss., Phase change material as a thermal energy storage material for cooling of building, Journal of Theoretical and Applied Information Technology,2005-2008,4(6),p.503-512, Available at http://www.jatit.org/volumes/research-papers/Vol4No6/6Vol4No6.pdf, (accessed10/05/2012).
- [3]. Ravikumar M., Srinivasan Dr.Pss, Natural cooling of building using phase change material, International Journal of Engineering and Technology, 2008, 5(1), p. 1-10, Available at: http://ijet.feiic.org/journals/J-2008-V1001.pdf, (accessed 10/05/2012).
- [4]. Zalba B., Marın J.M., Cabeza L.F., Mehling H., Review on thermal energy storage with phase change: materials, heat transfer analysis and applications, Applied Thermal Engineering 23, 2003, p.251–283.
- [5]. Patil N.D., Design and Analysis of Phase Change Material based thermal energy storage for active building cooling: a Review, International Journal of Engineering Science and Technology (IJEST),2012,4(6),p.2502-2509, Available at: http://www.ijest.info/docs/IJEST12-04-06-027.pdf, (accessed 10/06/2012).
- [6]. Khudhair A.M., Farid M.M., A review on energy conservation in building applications with thermal storage by latent heat using phase change materials, Energy Conversion and Management 2004, 45, p. 263-275.